ATTY.'S DOCKET: SUR-3645



IN THE UNI ED STATES PATENT AND TRADEMARK OFFICE

In re Application of:

Richard SMITH

Serial No.: 09/662,507

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For: SMALL-SCALE ! YDROGENOXIDIZING-DENITRIFYING
BIOREACTOR

Art Unit: 1724

Confirmation No. 2262

Washington D.C.

DECLARATION UNDER 37 CFR 1.131

I, Richard L. Smith, do hereby declare that I am the sole inventor of the above-captioned application.

Attached hereto are true copies of entries to my HOD Bioreactor Notebook #3, which entries were made prior to January ., 2000, the effective filing date of Rittmann et al., J.S. Patent No. 6.387,262. All work described in this declaration was conducted at the US Geological Survey in Boulder, Colorado.

It should be noted on page 80 that the notebook describes the four components of the apparatus as claimed, namely:

- a. autotroph: c, hydrogen-oxidizing denitrifying bacteria;
- a water electrolysis unit that provides a continual supply of oxygen-free hydrogen;
- c. a flow-though bioreactor that contains the HOD bacteria and is designed to maximize their ability to remove nitrate in the presence of hydrogen; and

d. a sand fill ration unit to remove unwanted microbial biomass from the treated water.

Additionally, opposite page 82 of the notebook pages is a figure of the hydrogen generator and denitrifying bioreactor and sand filter.

It is clear that this invention had been reduced to practice prior to January 1, 2000, because, as stated on page 84 of the notebook, optimum residence time had been determined.

I hereby further declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that all statements made on information and helief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 81 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon:

Richard T. Smith

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P. 05/07

Explanation of a specific embodiment of the Invention.

Theory of operation

The device described herein consists of 4 principle components. These are: 1) autotrophic, hydrogen-oxidizing denitrifying bacteria isolated from subsurface environments; 2) a low-oxst water electrohysis unit that provides a continual supply of oxygen-free hydrogen; 3) a flow-through bioreactor that contains the HOD bacteria and is designed to maximize their ebility to remove nitrate in the presence of hydrogen; and 4) a sand filtration unit to remove unwanted miloroblat blomass from the treated water.

carbonale, these bacteria can be used to remove nitrale in a water supply simply denitrification. Thus, oxygen must first be removed from any water supply before groundwaler are autotrophs (Smith et al., 94). That meens that they use carbon drinking water supply, as is hydrogen. In addition, many of the HOD bacteria in conditions. The HOD bacteria can also utilize hydrogen and respine aerobically. oxidizing hydrogen gas and coupling that to nitrate reduction (Figure 1). They by adding hydrogen gas. Such a treatment is very selective for HOD bacteria, occupy a unique ecclogical niche, one in which there is kille competition from This trait is very useful in a nitrate removal bioreactor because oxygen inhibits other microorganisms. The end products of the HOD process are water and Hydrogen-oxidizing denitrifying (HOD) bacteria obtain their energy by excluding all other types of microorganisms that could not grow under such culture can effect both oxygen and nitrate removat, as long as an adequate denitrification can commence within the reactor. However, the same HOD dioxide as a carbon source for growth; they have no additional carbon requirements. Because carbon dioxide is present in natural waters as Supply of hydrogen is available, Hydrogen gas has a low solubility in water. This low solubility requires that an excess of hydrogen is always available to remove the quantities of nitrate found in many contaminated water supplies. Hydrogen that is not utilized by acration. Hydrogen can be generated via electrolysis of water, which produces hydrogen gas at the snode and oxygen gas at the calhode at a molar stoichiometry of 2:1. The amount of hydrogen produced is dependent upon the vallage spylled to the electrodes and the electrolyte concentration.

Flow-through bioneactors are designed to provide a fixed stationary support for an attached nitcrobial biofilm. The biofilm contacts or is immersed in a flowing squeous stream and removes or alters the chemical composition of the water via the activity of the attached nitcroorganisms. In some cases, nutrients or substrates for the microorganisms need to be added to the bioneactor. If the substrate gas (such as hydrogen), counter current flow of the gas and the

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microorganisms. This can also serve as a mechanism to stilp other unwanted gases, such as oxygen, out of solution. water is advantageous to increase the availability of the gas to the

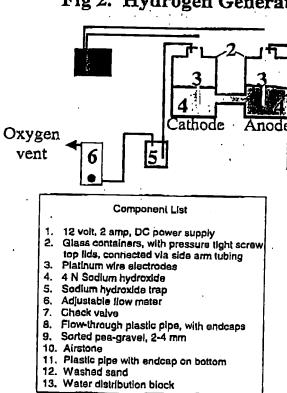
Detailed description of invention.

the text refer to the numbered items in the figures. consists of the 4 components listed in the above section. The numbers within scale water supply using the HOD reaction is shown in Figures 2 and 3 and An embodiment of the present invention to remove nitrate from a small

Component 1. HOD Bacteria.

Several strains of HOD bacteria have been isolated from groundwater and partially characterized. Strain HOD 5 is used in the present embodiment of this require no organic carbon for growth, only hydrogen, nitrale, and carbon dioxide which is identical to the function of the bioreactor. These microorganisms groundwater environments. This makes them ideal for such a treatment system this firvention. The bacteria have been Isolated from nitrate-containing bacleria are used as the reactive agents in the flow-through bioreactor used in erance un amine ex cabumbaixai fai iv walet nombi thombe à porous modium Pure cultures of autotrophic, hydrogen-oxidizing, dentiritying (HOD

Hydrogen Generator Fig 2.



Component 2. Hydrogen Generator.

Hydrogen gas is produced by hydrolysis of water in a dual-chamber, glass

hydrogen flowing, for 2-3 days before the water supply is turned on.

is transferred to the bioreactor column (see below; component 3) which has been

filled with HOD medium. The culture is grown statically in the bioreactor, with HOD medium (Smith et al. 1994). Following development of turbidity, the culture subclass of the proteobacteria and is most closely related to purple, non-sultur acetate, pyruvate, lactate, succinate and glutamate. Phylogenetic analysis of the

pure culture of HOD 5 is grown in batch culture on hydrogen and nitrate using phototrophic bacteria, particularly <u>Rhodocyclus</u> species. For the bioreactor, a full sequence of the 16S rRNA gene reveals that HOD 5 belongs to the beta

is a gram negative, motile rod, that can grow on hydrogen using either oxygen or

nitrate as electron acceptors. It can also grow aeroblcally on nutrient broth, invention. This strain is partially described in Smith et al (1994). The bacterium

valve (7) to prevent back flow, and into the bioreactor (8-10). Internal pressure anode chamber and is channeled through a sodium hydroxide trap (5), a check trap (5) to an adjustable gas flow controller (6). Hydrogen gas is produced in the cathode chamber, and is channeled via metal lubing through a sodium hydroxide commercial automobile battery charger (1). Oxygen gas is produced in line 2 amp DC electrical polential is continuously applied to the electrodes using a connected via hollow glass tubing and contain 4 N sodium hydroxide. A 12 vott cap that is penetrated with a platinum wire electrode (3). The chambers are reservoir (2). The two chambers are each sealed with a pressure-light screw top

> Hyerogen vent Nitrate-laden water 13 8 11 Q 12

Fig 3. Denitrifying Bioreactor

and Sand Filter

e Si

within the 2 chambers of the hydrogen generator is balanced using the adjustable flow controller.

Component 3. Flow-Through Biorestiggt.

The flow-through bioreactor (8-10) is constructed from plastic pipe and fitted with sealed endcaps. The bioreactor is filled with a coarse porous medium (9) such as washed pea gravel (2-4 mm diameter) or plastic or glass beads, which serve as solid surfaces to support biofilm formation by the HOD bacteria. Nitrate-laden water is pumped into the loo of the reactor, travels downward through the porous medium where it contacts the microbial biofilm, and exits out the bottom of the bloreactor nitrate-free. The water level within the bloreactor is controlled by the height of the exit tube.

Hydrogen gas enters the bioreactor via an airstone (10) in the bottom. Hydrogen bubbles travel upward, countercurrent to water flow, and are vented void to the countercurrent to water flow, and are vented void to the Hydrogen bubbles strip oxygen from the influent water and nitrogen gas norm water within the reactor that is produced via the denitrification reaction. The headspace votume in the bioreactor is designed not to exceed 1-5 % of the total votume of the bioreactor to minimize the amount of hydrogen gas present within the system.

Component 4. Sand Filtration Unit.

The nitrate-free water exiting the bioreactor then percolates via gravity flow through a sand filtration unit (11-13). This unit is constructed with plastic pipe that is fitted with a bottom endcap. The unit is filled with a bottom layer of pea gravel 4-6 inches thick, and overlain with clean, coarse- to medium-grained sand (12). On top of the sand column is a block (13) to evenly distribute the input water over the surface of the sand. The overall height of the sand filter unit is approximately equivalent to the height of the water column within the bioreactor. In the sand filter, the water is serated and filtered to remove suspended microorganisms from the bloreactor effluent. The top layer of sand within the infiltration unit is periodically removed and replaced with cleen sand. Water exits the sand filter unit via a tube inserted in the bottom endcap.

3. Preferred and extreme ranges of conditions.

Optimum hydraulic residence time for the bloreactor for a nitrate concentration of 2 mM (28 mg/L N) is 1.5.2 hours at a temperature of 25 °C. The bloreactor can effectively remove nitrate concentrations from 0.7-20 mM (10-280 mg/L N) in a pH range of 6-9.